



(11) **EP 0 997 498 A1**

(12) **EUROPEAN PATENT APPLICATION**

(43) Date of publication:
03.05.2000 Bulletin 2000/18

(51) Int. Cl.⁷: C08L 83/04, C08L 83/07,
H01L 23/29

(21) Application number: 99308319.5

(22) Date of filing: 21.10.1999

(84) Designated Contracting States:
AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU
MC NL PT SE
Designated Extension States:
AL LT LV MK RO SI

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(30) Priority: 28.10.1998 US 181211

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(54) **Silicone composition, method for the preparation thereof and silicone elastomer**

(57) A silicone composition comprising (A) 100 parts by weight of a polydiorganosiloxane containing an average of at least two silicon-bonded alkenyl groups per molecule; (B) 75 to 150 parts by weight of an organopolysiloxane resin having a number-average molecular weight of from 2,000 to 5,000 and comprising $R^3_2R^4SiO_{1/2}$ units and $SiO_{4/2}$ units wherein each R^3 is independently selected from monovalent hydrocarbon and monovalent halogenated hydrocarbon groups free of aliphatic unsaturation, R^4 is selected from R^3 and alkenyl, the mole ratio of $R^3_2R^4SiO_{1/2}$ units to $SiO_{4/2}$ units is from 0.6:1 to 1.1:1, and the resin contains an average of from 2.5 to 7.5 mole percent of alkenyl groups; (C) an organohydrogenpolysiloxane having an average of at least three silicon-bonded hydrogen atoms per molecule in an amount to provide from one to three silicon-bonded hydrogen atoms per alkenyl group in components (A) and (B) combined; (D) an adhesion promoter in an amount to effect adhesion of the composition to a substrate; and (E) a hydrosilylation catalyst in an amount to cure the composition.

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Description

[0001] The present invention relates to a silicone composition and more particularly to an addition-curable silicone composition containing a critical concentration of an organopolysiloxane resin having a defined molecular weight. This invention also relates to a method for the preparation of this composition and to a silicone elastomer produced therefrom.

[0002] Silicones are widely employed in the electrical and electronics industries due to their unique properties, which include low alpha particle emissions, very good moisture resistance, excellent electrical insulation, excellent thermal stability and very high ionic purity. In particular, silicone encapsulants are used to improve the reliability of electronic devices by providing protection against environmental moisture, UV radiation, ozone and weathering. Moreover, recent advances in semiconductor packaging, namely, the development of chip scale or chip size packages, have created a critical demand for high performance silicone encapsulants.

[0003] Curable silicone compositions comprising a polydimethylsiloxane polymer and an organopolysiloxane resin are well known in the art. These compositions typically employ up to 50 parts by weight of resin per 100 parts by weight of polymer. Furthermore, several cure systems are recognized, including hydrosilylation, peroxide, moisture, condensation and radiation.

[0004] However, silicone elastomers produced from such compositions typically exhibit several fold increases in coefficient of thermal expansion and modulus in the temperature range corresponding to the amorphous-crystalline phase transition, typically from -50 to -40°C., rendering them unsuitable for use as encapsulants in chip scale packages at very low temperatures. As the amorphous-crystalline transition is traversed during thermal cycling, stresses develop in the chip scale package due to thermal expansion and contraction of the silicone encapsulant and the difference in coefficients of thermal expansion between the substrate and silicon die. Thermally-induced stresses can damage critical components of a semiconductor package, including solder joints and lead wires, resulting in poor reliability.

[0005] Inorganic fillers such as silica and alumina are typically added to silicone compositions to reduce the coefficient of thermal expansion of cured products. Such fillers have very low coefficients of thermal expansion compared to a silicone elastomer and thus produce an overall reduction in thermal expansion of the elastomer. However, fillers also increase the viscosity of a silicone composition and, consequently, often reduce processability.

[0006] Various approaches to controlling the moduli of silicone compositions have been reported in the literature. For example, nonfunctional silicone oils have been employed to regulate the moduli of elastomers produced from curable silicone compositions. Unfortunately, silicone oils have a tendency to undergo post cure migration, causing shrinkage of the composition and contamination of surrounding materials.

[0007] Reactive silicone oils have also been used in silicone compositions to lower the modulus of a cured silicone rubber. For example, U.S. Patent 5,216,104 discloses an addition-type curable silicone rubber composition comprising (A) an organopolysiloxane containing at least two alkenyl groups per molecule, (B) an organopolysiloxane containing an average of one alkenyl group per molecule, (C) an organohydrogenpolysiloxane containing at least two silicon-bonded hydrogen atoms per molecule, (D) a platinum catalyst, and optionally an organopolysiloxane resin. The '104 patent teaches that the silicone rubber composition can give a low modulus silicone rubber by curing which neither contaminates the material in contact therewith nor deforms owing to shrinkage, without hindering the adhesion of the composition during its curing stage. Unlike the '104 patent, the present invention does not require an organopolysiloxane containing an average of one alkenyl group per molecule. Also, the '104 patent does not teach the molecular weight limitations on the organopolysiloxane resin, resin to polymer ratio and adhesion promoter of the present invention.

[0008] Organopolysiloxane resins have also been used in silicone compositions to regulate the moduli of cured materials. For example, U.S. Patent 5,705,587 teaches a room temperature-curable silicone elastomer composition comprising (A) a diorganopolysiloxane composed of (a) a silanol-enclosed diorganopolysiloxane and (b) a diorganopolysiloxane that is endblocked at one molecular chain terminal by silanol and at the other terminal by trialkylsiloxy; (B) an alkoxy functional silane; (C) an organopolysiloxane resin; (D) a curing catalyst; and (E) a diorganopolysiloxane containing at least two carboxyl groups. The '587 patent also teaches that the aforementioned composition has an excellent pre-cure workability and cures into a low-modulus, high-elongation silicone elastomer. However, the '587 patent does not teach the polydiorganopolysiloxane polymer, organopolysiloxane resin, resin to polymer ratio, organohydrogenpolysiloxane and hydrosilylation catalyst of the present invention.

[0009] U.S. Patent 5,373,078 discloses curable organosiloxane compositions comprising a liquid polyorganosiloxane having an average of at least two alkenyl groups per molecule, (B) a liquid organohydrogenpolysiloxane having at least three silicon-bonded hydrogen atoms per molecule, (C) a hydrosilylation catalyst, and (D) a liquid MQ organosiloxane resin. The '078 patent also teaches that the liquid MQ resin reduces the viscosity of the curable compositions and the moduli of cured elastomers prepared from the compositions without any substantial adverse effect on the physical properties of the cured elastomers. However, the '078 patent does not teach the organopolysiloxane resin, resin to polymer ratio and adhesion promoter of the present invention.

[0010] EP 0767216 discloses moisture-cure and addition-cure silicone resin/fluid alloy compositions that exhibit

high strength and toughness over a predetermined service temperature range comprising (A) an organosilicone resin, (B) a predominately linear silicone fluid, and (C) a predetermined quantity of a crosslinker having at least three functional groups per molecule. The EP 0767216 patent teaches that the crosslinked product of (A) and (B), when free of an inorganic filler, exhibit a shear loss modulus (G'') greater than or equal to 10^5 Pa and a loss tangent greater than or equal to 0.10. However, the EP0767216 application further discloses that such compositions have utility in the manufacture of molded and extruded articles and, therefore, does not teach the adhesion promoter of the present invention.

[0011] U.S. Patent 5,756,598 discloses an organosiloxane composition curable to a silicone elastomer comprising (A) a polydiorganosiloxane having an average of at least two silicon-bonded alkenyl groups per molecule; (B) an organohydrogenpolysiloxane having an average of at least two silicon-bonded hydrogen atoms per molecule; (C) a platinum group catalyst, and (D) a resinous organosiloxane copolymer having an average of at least two alkenyl groups per molecule. The '598 patent also teaches that the aforementioned composition is curable to a silicone elastomer having improved weep properties. However, the '598 patent does not teach the resin to polymer ratio of the present invention. [0012] The present inventors have discovered that a silicone composition comprising a critical concentration of an organopolysiloxane resin having a number-average molecular weight within a specified range cures to form an elastomer having excellent adhesion to a variety of substrates and a very low coefficient of thermal expansion and storage modulus over a wide temperature range.

[0013] The present invention is directed to a silicone composition comprising:

(A) 100 parts by weight of a polydiorganosiloxane containing an average of at least two silicon-bonded alkenyl groups per molecule;

(B) 75 to 150 parts by weight of an organopolysiloxane resin having a number-average molecular weight of from 2,000 to 5,000 and comprising $R^3_2R^4SiO_{1/2}$ siloxane units and $SiO_{4/2}$ siloxane units wherein each R^3 is independently selected from monovalent hydrocarbon and monovalent halogenated hydrocarbon groups free of aliphatic unsaturation, R^4 is selected from R^3 and alkenyl groups, the mole ratio of $R^3_2R^4SiO_{1/2}$ units to $SiO_{4/2}$ units is from 0.6:1 to 1.1:1, and the resin contains an average of from 2.5 to 7.5 mole percent of alkenyl groups;

(C) an organohydrogenpolysiloxane having an average of at least three silicon-bonded hydrogen atoms per molecule in an amount to provide from one to three silicon-bonded hydrogen atoms per alkenyl group in components (A) and (B) combined;

(D) an adhesion promoter in an amount to effect adhesion of the composition to a substrate; and

(E) a hydrosilylation catalyst in an amount to cure the composition.

[0014] The present invention is also directed to a silicone elastomer comprising a reaction product of the composition described above.

[0015] The present invention is further directed to a multi-part silicone composition comprising components (A) through (E) specified above, with the proviso that neither the polydiorganosiloxane nor the organopolysiloxane resin are present with the organohydrogenpolysiloxane and the hydrosilylation catalyst in the same part.

[0016] A method for preparing the silicone composition of this invention comprises the steps of mixing components (A) through (E).

[0017] The invention is still further directed to a chip scale package comprising a semiconductor chip, a supporting substrate, at least one member that connects the semiconductor chip to the substrate and the silicone composition described above.

[0018] The silicone composition of the present invention possesses the rheological properties required for a number of applications. For example the composition of the instant invention exhibits rapid flow around and/or under a silicon chip or die and is particularly suitable for use in high throughput encapsulation processes. In particular, the viscosity of the composition is sufficiently low to ensure rapid and complete filling of the semiconductor device and to allow escape of entrapped air bubbles. It is not necessary to dilute the composition of the present invention with an organic solvent to achieve a workable viscosity. Consequently, the present solventless composition avoids the health and environmental hazards associated with solvent-borne silicones. Furthermore, during curing, the solventless composition typically undergoes less shrinkage than solvent-borne silicone compositions.

[0019] The cured silicone composition of the present invention also displays excellent adhesion to a wide variety of materials, including those commonly employed in chip scale packages, such as metals, glass, silicon, silicon dioxide, ceramics, rubbers and plastics. In particular, the cured composition has excellent adhesion to flexible support materials used in chip scale packages, particularly polyimide tape.

[0020] Moreover, the silicone composition of the present invention cures to form an elastomer having excellent thermal stability and flexibility in the temperature range from -65 to 150°C, as evidenced by low thermal expansion and low modulus. In particular, the cured composition of the present invention has a coefficient of thermal expansion typically less than 300 $\mu\text{m/m}^\circ\text{C}$ in the temperature range from -50 to -40°C. Also, the ratio of the storage modulus at -65°C to the storage modulus at 20°C is typically less than 5. As a consequence of low thermal expansion and low modulus, the

elastomer of the instant invention exhibits minimal dimensional change and excellent flexibility at low temperatures. The flexible silicone elastomer relieves thermally induced stress on delicate leads and solder connections in a semiconductor package due to a difference in the coefficients of thermal expansion between the silicon die and substrate.

[0021] Additionally, the present silicone composition has high radiological purity (low alpha particle emissions) required for RAM applications; low levels of ionic impurities (e.g., sodium, potassium, chloride); and excellent electrical properties such as a low dielectric constant and dissipation factor. The composition of the present invention is particularly useful as an encapsulant in chip scale or chip size semiconductor packages.

[0022] Component (A) of the present invention is a polydiorganosiloxane containing an average of at least two silicon-bonded alkenyl groups per molecule. Suitable alkenyl groups contain from 2 to 6 carbon atoms and are exemplified by, but not limited to vinyl, allyl and hexenyl. The alkenyl groups in component (A) may be located at terminal, pendant, or both terminal and pendant positions. The remaining silicon-bonded organic groups in component (A) are independently selected from monovalent hydrocarbon and monovalent halogenated hydrocarbon groups free of aliphatic unsaturation. These groups typically contain from 1 to 8 carbon atoms, preferably from 1 to 4 carbon atoms, and are exemplified by, but not limited to alkyl such as methyl, ethyl, propyl and butyl; aryl such as phenyl; and halogenated alkyl such as 3,3,3-trifluoropropyl. Typically, at least 50 percent of the organic groups in component (A) are methyl.

[0023] The structure of component (A) is typically linear, however it may contain some branching due to the presence of trifunctional siloxane units. The viscosity of component (A) at 25°C., which varies with molecular weight and structure, is typically from 0.5 to 100 Pa · s, preferably from 1 to 50 Pa · s, and more preferably from 2 to 30 Pa · s.

[0024] Preferably, component (A) is a polydiorganosiloxane having the general formula $R^1R^2SiO(R^1)_2SiO(R^1)_2SiR^2$ wherein each R^1 is independently selected from monovalent hydrocarbon and monovalent halogenated hydrocarbon groups free of aliphatic unsaturation, as defined above; R^2 is alkenyl, as defined above; and n has a value such that the viscosity of component (A) at 25°C. is in one of the ranges cited above. Preferably, R^1 is methyl and R^2 is vinyl.

[0025] Component (A) can comprise a single polydiorganosiloxane or a mixture of two or more polydiorganosiloxanes having different viscosities. For example, component (A) can comprise a mixture of a first polydiorganosiloxane having a viscosity of from 40 to 100 Pa · s at 25°C. and a second polydiorganosiloxane having a viscosity of from 1 to 10 Pa · s at 25°C. In a preferred embodiment, component (A) comprises a first polydiorganosiloxane having a viscosity of from 45 to 65 Pa · s at 25°C., and 10 to 50 percent, preferably 25 to 35 percent by weight, of a polydiorganosiloxane having a viscosity of from 1.8 to 2.4 Pa · s at 25°C. In this embodiment, the lower viscosity polydiorganosiloxane improves both the flow of the liquid composition and the solvent resistance of the cured silicone product.

[0026] Specific examples of polydiorganosiloxanes useful in the present invention include, but are not limited to, the following:

$ViMe_2SiO(Me_2SiO)_nSiMe_2Vi$, $ViMe_2SiO(Me_2SiO)_{0.92n}(MePhSiO)_{0.08n}SiMe_2Vi$,
 $ViMe_2SiO(Me_2SiO)_{0.98n}(MeViSiO)_{0.02n}SiMe_2Vi$, $ViMePhSiO(Me_2SiO)_{0.92n}(MePhSiO)_{0.08n}SiMePhVi$,
 $Me_2SiO(Me_2SiO)_{0.95n}(MeViSiO)_{0.05n}SiMe_3$, $PhMeViSiO(Me_2SiO)_nSiPhMeVi$, and others, where Me, Vi and Ph denote methyl, vinyl and phenyl respectively and n is as defined above. Preferred polydiorganosiloxanes are dimethylvinylsiloxy-terminated polydimethylsiloxanes.

[0027] Methods for preparing component (A) of the present composition, such as hydrolysis and condensation of the corresponding halosilanes or equilibration of cyclic polydiorganosiloxanes, are well known in the art.

[0028] Component (B) of the present invention is an organopolysiloxane resin consisting essentially of $R^3R^4SiO_{1/2}$ siloxane units and $SiO_{4/2}$ siloxane units wherein each R^3 is independently selected from monovalent hydrocarbon and monovalent halogenated hydrocarbon groups free of aliphatic unsaturation, and R^4 is selected from R^3 and alkenyl. The monovalent hydrocarbon and monovalent halogenated hydrocarbon groups free of aliphatic unsaturation and alkenyl groups in component (B) are identical to those defined above for component (A). Typically at least 50 percent, and preferably all of the R^3 groups are methyl. The preferred alkenyl group for R^4 is vinyl.

[0029] The mole ratio of $R^3R^4SiO_{1/2}$ units to $SiO_{4/2}$ units is from 0.6 to 1.1, and preferably from 0.9 to 1.1, as determined by ^{29}Si nuclear magnetic resonance ($^{29}SiNMR$) spectrometry. The M/Q ratio represents the total number of M units to the total number of Q units in component (B) and includes contributions from any neopentamer present, described below.

[0030] Component (B) contains $HOSiO_{3/2}$ units (TOH units), which account for the silicon-bonded hydroxyl content of the organopolysiloxane resin. The silicon-bonded hydroxyl content of component (B), as determined by $^{29}SiNMR$ spectrometry, is typically less than 2 percent by weight and preferably less than 1 percent by weight, based on the total weight of the resin.

[0031] Component (B) contains from 2.5 to 7.5 mole percent, and more preferably from 4.0 to 6.5 mole percent, of alkenyl groups. The mole percent of alkenyl groups in the resin is defined here as the ratio of the number of moles of alkenyl-containing siloxane units in the resin to the total number of moles of siloxane units in the resin, multiplied by 100. The total number of moles of siloxane units in the resin includes the M, Q and TOH units described above. When the alkenyl content of the resin is less than 2.5 mole percent, the cured composition becomes soft and tacky and the ratio of the storage modulus at -65°C. to the storage modulus at 20°C. increases. When the alkenyl content of the resin is

greater than 7.5 mole percent, the cured composition becomes hard and brittle, as evidenced by an increase in storage modulus, G'.

[0032] Component (B) can comprise a single organopolysiloxane resin or a mixture of two or more organopolysiloxane resins. For example, component (B) can comprise a mixture of a first resin containing 6.0 mole percent of vinyl groups and 50 percent by weight of a resin free of vinyl groups. Such organopolysiloxane would contain an average of 3.0 mole percent of vinyl groups, thus meeting the required alkenyl content for component (A).

[0033] Component (B) can also contain a small amount of a low molecular weight material comprised substantially of a neopentamer organopolysiloxane having the formula $(R_3SiO_{1/2})_5Si$, the latter material being a byproduct in the preparation of the resin according to the method of Dault et al., described infra. Additionally, component (B) may contain minor amounts of diorganosiloxane.

[0034] Component (B) has a number-average molecular weight (M_n) of from 2,000 to 5,000, preferably from 2,000 to 4,000, and more preferably from 2,000 to 3,000 when measured by gel permeation chromatography (GPC), the neopentamer peak being excluded from the measurement. In this molecular weight determination, narrow fractions of MQ resins are used to calibrate the GPC equipment, the absolute molecular weights of the fractions being first ascertained by a technique such as vapor phase osmometry.

[0035] A preferred organopolysiloxane resin has a number-average molecular weight of 2,500 and comprises $(CH_3)_2CH_2=CHSiO_{1/2}$ siloxane units, $(CH_3)_3SiO_{1/2}$ siloxane units, and $SiO_{4/2}$ siloxane units, wherein the mole ratio of the combination of $(CH_3)_2CH_2=CHSiO_{1/2}$ units and $(CH_3)_3SiO_{1/2}$ units to $SiO_{4/2}$ units is 1.0, and the resin contains 6.0 mole percent of vinyl groups.

[0036] The concentration of component (B) in the composition of the present invention is critical with respect to the pre-cure viscosity and the properties achieved following curing, particularly thermal expansion and modulus. The concentration of component (B) is from 75 to 150 parts by weight and preferably from 90 to 125 parts by weight per 100 parts by weight of component (A). When the concentration of component (B) is less than 75 parts by weight, the coefficient of thermal expansion in the range from -50 to -40°C. and the storage modulus of the silicone elastomer at -65°C. increase significantly relative to their values at 20°C. At concentrations greater than 150 parts by weight, the viscosity of the uncured composition increases to the extent that it does not flow at a rate useful for commercial encapsulation processes.

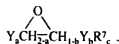
[0037] Component (B) of the present invention can be prepared by treating a resin copolymer produced by the silica hydrosol capping process of U.S. Patent 2,676,182, as modified by U.S. Patent 3,627,851 and U.S. Patent 3,772,247, with an alkenyl-containing endblocking agent.

[0038] Briefly stated, the modified method of Dault et al. used to prepare the organopolysiloxane resin of the present invention comprises limiting the concentration of the sodium silicate solution and/or the silicon-to-sodium ratio in the sodium silicate and/or the time before capping the neutralized sodium silicate solution to generally lower values than those disclosed by Dault et al. in order to prevent excessive growth of the silica particles and to obtain an organopolysiloxane having the preferred M_n . Thus, the preferred silicate concentration is generally limited to a value of 40 to 120, preferably 60 to 100, and most preferably 75 grams/liter. The neutralized silica hydrosol is preferably stabilized with an alcohol, such as isopropanol and capped with $(CH_3)_3SiO_{1/2}$ siloxane units as soon as possible, preferably within 30 seconds after being neutralized. The sodium silicate employed preferably has the formula $Na_2O \cdot xSiO_2$, wherein x has a value of from 2 to 3.5. The resulting copolymers contain from 2 to 3 percent by weight of hydroxyl groups.

[0039] The resin of the present invention, which contains less than 2 percent by weight of silicon-bonded hydroxyl groups, can be prepared by reacting the product of the modified method of Dault et al. with an alkenyl-containing silazane, siloxane or silane in an amount to provide from 2.5 to 7.5 mole percent of alkenyl groups in the final product. Alkenyl-containing endblocking agents are known in the art and exemplified in U.S. Patents 4,584,355, 4,591,622 and 4,585,836.

[0040] It will be understood that as the viscosity of component (A) increases in the range from 5 to 100 Pa · s, it may be necessary to decrease the concentration of the organopolysiloxane resin within the above-stated range to obtain a processable composition. Also, as the alkenyl content of the organopolysiloxane resin increases in the range from 2.5 to 7.5 mole percent, it may be necessary to reduce the concentration of the resin to obtain a silicone elastomer having satisfactory adhesion.

[0041] Component (C) of the present invention is an organohydrogenpolysiloxane having an average of at least three silicon-bonded hydrogen atoms per molecule and an average of no more than one silicon-bonded hydrogen atom per silicon atom. The silicon-bonded hydrogen atoms can be located at terminal, pendant, or at both terminal and pendant positions in the organohydrogenpolysiloxane. Component (C) can be a homopolymer or a copolymer. The structure of the organohydrogenpolysiloxane can be linear, branched or cyclic. The siloxane units present in component (C) may include $HR^3SiO_{1/2}$, $R^3SiO_{1/2}$, $HR^3SiO_{2/2}$, $R^3SiO_{2/2}$, $HSiO_{3/2}$, $R^3SiO_{3/2}$ and $SiO_{4/2}$ units. In the preceding formulae each R^3 is independently selected from monovalent hydrocarbon and monovalent halogenated hydrocarbon groups free of aliphatic unsaturation, as defined previously for component (A). Typically at least 50 percent, and more preferably substantially all of the organic groups in component (C) are methyl. Component (C) can be a single organohydro-



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wherein each Y is independently an alkyl group having 1 or 2 carbon atoms; a is 0, 1, or 2; b and c are each 0 or 1; and R⁷ is a divalent hydrocarbon group having no more than 12 carbon atoms. Preferably, R⁷ is selected from a saturated aliphatic hydrocarbon group, an arylene group, and a divalent group having the formula -R⁸(OR⁹)_dOR⁸, wherein R⁸ is a divalent saturated aliphatic hydrocarbon group having 2 to 6 carbon atoms and d has a value of from 0 to 8.

[0051] The remaining silicon-bonded organic groups in the epoxy-containing alkoxy silane are independently selected from monovalent hydrocarbon groups having less than 7 carbon atoms and monovalent fluorinated alkyl groups having less than 7 carbon atoms. The monovalent hydrocarbon groups are exemplified by alkyl, such as methyl, ethyl, propyl, hexyl; alkenyl, such as vinyl; and aryl, such as phenyl. Examples of suitable fluorinated alkyl groups include perfluoroalkyl, such as 3,3,3-trifluoropropyl, β-(perfluoroethyl)ethyl and β-(perfluoropropyl)ethyl.

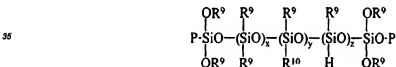
[0052] The structure of epoxy-containing alkoxy silane (ii) can be linear, branched or cyclic. Silane (ii) can be a single silane or a mixture of two or more different silanes. Preferably, epoxy-containing alkoxy silane (ii) is a mono(epoxy-organo)trialkoxysilane. A particularly preferred epoxy-containing alkoxy silane is glycidoxopropyltrimethoxysilane.

Methods for the preparation of such silanes are well known in the art.

[0053] The two components of the aforementioned adhesion promoter can be either added separately or as a blend to the composition of the present invention. Such compositions also typically contain up to 10 percent by weight of at least one reaction product of the hydroxy-containing polysiloxane and the alkoxy silane, which is normally formed during storage under ambient conditions.

[0054] A specific example of a preferred adhesion promoter according to the present invention is a blend comprising 43.5 percent by weight of a hydroxyl-terminated polymethylvinylsiloxane containing 2.25 to 4 percent by weight of silicon-bonded hydroxyl groups and 27.5 percent by weight of vinyl groups, 50 percent by weight of glycidoxopropyltrimethoxysilane, and 6.5 percent by weight of a reaction product of the hydroxyl-terminated polymethylvinylsiloxane and glycidoxopropyltrimethoxysilane. Adhesion promoters are disclosed in U.S. Patent 4,087,585.

[0055] Preferred adhesion promoters according to the present invention also include organosilicon compounds having the average formula:



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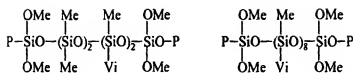
wherein R⁹ is a monovalent hydrocarbon group free of aliphatic unsaturation, R¹⁰ is alkenyl, x has a value of from 0 to 10, y has a value of from 0 to 10, z has a value of 0 to 10, the sum y + z is at least 1, and P has the formula:



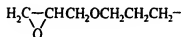
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wherein R¹¹ is a divalent organic group. Suitable R⁹ groups generally contain from 1 to 6 carbon atoms and are exemplified by methyl, ethyl, propyl and butyl. Preferably, R⁹ is methyl. The R¹⁰ groups typically have from 2 to 6 carbon atoms and include vinyl, allyl, 1-propenyl and isopropenyl. Preferably, R¹⁰ is vinyl. Examples of R¹¹ groups include methylene, ethylene, propylene, phenylene, chloroethylene, fluoroethylene, -CH₂OCH₂CH₂CH₂-, -CH₂CH₂OCH₂CH₂-, -CH₂CH₂OCH(CH₃)CH₂-, and -CH₂OCH₂CH₂OCH₂CH₂-. Preferably, R¹¹ is -CH₂OCH₂CH₂CH₂-. The preceding class of adhesion promoters, inter alia, is exemplified in U.S. Patent 4,082,726. Methods for the preparation of these organosilicon compounds from hydroxy-terminated diorganosiloxanes and epoxy-containing alkoxy silanes are well known in the art.

[0056] Specific examples of such preferred adhesion promoters according to the present invention are



and mixtures thereof, wherein P is



[0067] Component (E) of the present invention is a hydrosilylation catalyst comprising a platinum group metal or a compound containing such a metal that promotes the addition reaction of components (A) and (B) with component (C). These metals include platinum, rhodium, ruthenium, palladium, osmium and iridium. Platinum and platinum compounds are preferred based on the high activity level of these catalysts in hydrosilylation reactions. A preferred class of platinum catalysts are the complexes of chloroplatinic acid with certain vinyl-containing organosiloxane compounds disclosed by Willing in U.S. Patent 3,419,593. A particularly preferred catalyst of this type is the reaction product of chloroplatinic acid and 1,3-diethenyl-1,1,3,3-tetramethyldisiloxane.

[0058] The hydrosilylation catalyst is present in an amount sufficient to cure the composition of the present invention. Typically, the concentration of component (E) is enough to provide from 0.1 to 1000, preferably from 1 to 500, and more preferably from 1 to 50 parts by weight of a platinum group metal, per one million parts by weight of the combined weight of components (A), (B) and (C). The rate of cure is very slow below 0.1 ppm of platinum group metal. The use of more than 1000 ppm of platinum group metal results in no appreciable increase in cure rate and is therefore uneconomical.

[0069] Mixtures of the components (A), (B), (C), (D) and (E) may begin to cure at ambient temperature. To obtain a longer working time or "pot life", the activity of the catalyst under ambient conditions can be retarded or suppressed by the addition of a suitable inhibitor to the composition of the present invention. A platinum catalyst inhibitor retards curing of the present composition at ambient temperature, but does not prevent the composition from curing at elevated temperatures. In order to be effective in this invention, the platinum catalyst inhibitor must be soluble in the composition. Suitable platinum catalyst inhibitors include various "ene-yne" systems such as 3-methyl-3-penten-1-yne and 3,5-dimethyl-3-hexen-1-yne; acetylenic alcohols such as 3,5-dimethyl-1-hexyn-3-ol, 1-ethynyl-1-cyclohexanol and 2-phenyl-3-butyne-2-ol; maleates and fumarates, such as the well known dialkyl, dialkenyl, and dialkoxyalkyl fumarates and maleates; cyclovinylsiloxanes; and benzyl alcohol.

[0060] Acetylenic alcohols constitute a preferred class of inhibitors due to their high activity in hydrosilylation reactions. 3,5-Dimethyl-1-hexyn-3-ol is a particularly preferred inhibitor in the composition of the present invention. Compositions containing these inhibitors generally require heating at 70°C. or above to cure at a practical rate.

[0061] The concentration of platinum catalyst inhibitor in the present composition is sufficient to retard curing of the composition at ambient temperature without preventing or excessively prolonging cure at elevated temperatures. This concentration will vary widely depending on the particular inhibitor used, the nature and concentration of the hydrosilylation catalyst and the nature of the organohydrogenpolyisiloxane.

[0062] Inhibitor concentrations as low as one mole of inhibitor per mole of platinum group metal will in some instances yield a satisfactory storage stability and cure rate. In other instances, inhibitor concentrations of up to 500 or more moles of inhibitor per mole of platinum group metal may be required. Generally, the inhibitor is present in an amount from 1 to 100 moles per mole of platinum group metal. The optimum concentration for a particular inhibitor in a given composition can be readily determined by routine experimentation.

[0063] The composition of the present invention may further comprise an inorganic filler. Fillers suitable for use in the composition of the present invention have a high radiological purity; low coefficient of thermal expansion; low levels of ionic impurities such as sodium, potassium, and chloride; and low moisture content. Examples of preferred fillers include fused silica (fused quartz), alumina, boron nitride and aluminum nitride. Fused silica is a particularly preferred filler in compositions used to encapsulate random access memory (RAM) devices, which are very sensitive to alpha particle emissions.

[0064] The average particle size of the filler is typically from 2 to 25 μm and preferably from 2 to 10 μm . When the average particle size is less than 2 μm , the viscosity of the composition may be too high for use in conventional encapsulation processes. When the average particle size is greater than 25 μm , the particles may be excluded by the comparatively smaller dimensions of a semiconductor device. Also, the larger filler particles tend to settle in the composition rather than remaining in suspension.

[0065] Although, the shape of the filler particles is not critical, particles having a spherical shape are preferred because they generally impart a smaller increase in viscosity to the composition than particles having other shapes.

[0066] The concentration of filler in the present composition is typically from 10 to 300 parts by weight per 100 parts by weight of component (A). However, the filler should not be used in an amount that increases the viscosity of the composition above 60 Pa \cdot s at 25°C, prior to cure, the viscosity measured using a controlled stress viscometer at a shear rate of 0.1 sec^{-1} . Above the specified viscosity, the composition is too viscous for practical use in conventional encapsulation processes. It will be understood that as the concentration of the inorganic filler increases in the range from 10 to 300 parts by weight per 100 parts by weight of component (A), it may be necessary to reduce the concentration of component (B) within the aforementioned range to achieve a processable composition.

[0067] The composition of the instant invention is typically prepared by combining components (A) through (E) and, optionally a filler and/or inhibitor, in the stated proportions. Mixing can be accomplished by any of the techniques known in the art such as milling, blending, and stirring, either in a batch or continuous process. The particular device is determined by the viscosity of the components and the final composition. Preferably, the hydrosilylation catalyst is added last at a temperature below 30°C, to prevent premature curing of the composition and thus ensure adequate working time.

[0068] Alternatively, the composition of the present invention can be a multi-part composition comprising components (A) through (E) in two or more parts. The multi-part composition can contain any number of parts, provided that neither the polydiorganosiloxane nor the organopolysiloxane resin are present with the organohydrogenopolysiloxane and hydrosilylation catalyst in the same part. In a typical method for preparing such a composition, a portion of the polydiorganosiloxane, a portion of the organopolysiloxane resin, the adhesion promoter, the hydrosilylation catalyst, and any filler or additives are mixed together to produce Part A, and the remaining portions of the polydiorganosiloxane and resin, organohydrogenopolysiloxane and platinum catalyst inhibitor are mixed together to produce Part B. Preferably, the components are packaged in such a manner that equal weight amounts of each package can be mixed to produce the one part composition of this invention.

[0069] The compositions of the present invention should be stored in sealed containers to prevent exposure to air and moisture. The one part composition of the present invention may be stored at room temperature for several weeks without any change in the properties of the cured encapsulant product. However, the shelf life of the composition of this invention can be extended to several months by storing the mixtures at a temperature below 0°C, and preferably from -30 to -20°C. Individual sealed packages of the multi-part composition described above can be stored for over 6 months at ambient conditions without any deterioration in the performance of the composition produced upon their admixture.

[0070] The composition of this invention can be cured by heating at temperatures of from 70 to 200°C, preferably from 100 to 150°C, for a suitable length of time. For example, the composition typically cures in one hour at 150°C.

[0071] In general, the composition of the present invention can further comprise small amounts of additional ingredients typically used in the formulation of silicone compositions, such as antioxidants, pigments, stabilizers, and fillers, provided the ingredient does not adversely affect the physical properties of the cured product, particularly the coefficient of thermal expansion and the storage modulus. Carbon black is a preferred pigment in the present composition.

[0072] The composition of the present invention is particularly useful as an encapsulant in chip scale packages. As used herein the term "chip scale" conforms to the definition proposed by the Electronic Industries Association and the Institute for Interconnecting and Packaging Electronic Circuits in standard J-Stid-012, "Implementation of Flip Chip and Chip Scale Technology," January 1996, p. 1. Section 1.2 of the document defines chip scale technology as follows:

"Chip scale technology is categorized as semiconductor chip structures that have been made robust to facilitate ease of chip handling, testing, and chip assembly. The chip scale technologies have common attributes of minimal size, no more than 1.2X the area of the original die size, and are direct surface mountable."

[0073] The chip scale package of the present invention comprises a semiconductor chip, a supporting substrate, at least one member connecting the semiconductor chip to the substrate, and the silicone composition described above. There are no restrictions on the semiconductor chip employed in the chip scale package of the present invention. The supporting substrate can be substantially rigid, semi-rigid, or flexible. However, typically the supporting substrate is comprised of a flexible dielectric film. Preferably, the dielectric film is a polymeric material such as polyimide. The member connecting the chip to the substrate typically comprises wire leads and solder connections.

[0074] Chip scale packages according to the present invention are exemplified by, but not limited to, the flexible interposer type chip scale packages disclosed by T. Chung in "Updates on Worldwide Chip Scale Packaging (II)". U.S. Patents 5,659,952 and 5,663,106 disclose preferred chip scale packages, which include an encapsulation material.

[0075] The following examples are presented to further illustrate the composition of this invention, but are not to be considered as limiting the invention, which is delineated in the appended claims. All parts and percentages reported in

the examples are by weight. The following methods and materials were employed in the examples:

- [0076] Cured silicone compositions having a thickness of 1.9 mm were prepared by compression molding a curable composition between two Teflon-lined aluminum plates under a pressure of 10,000 Kg at room temperature for two minutes. The plates were transferred to a hot press and maintained under a pressure of 10,000 Kg at 150°C. for fifteen minutes. The plates were immediately removed from the hot press and placed in an oven at 150°C. for 1 hour. After allowing the plates to cool to room temperature, the cured composition was carefully removed from the die assembly.
- [0077] The number-average molecular weight (M_n) of an organopolysiloxane resin were determined by gel permeation chromatography (GPC) using Varian TSK 4000+2500 columns at 35°C., a chloroform mobile phase at 1 mL/min and a refractive index detector. Narrow fractions of MQ resins were used to calibrate the GPC equipment.
- [0078] The viscosity of a polymer or curable composition was determined using a Rheometrics SR-5000 controlled-stress rheometer equipped with parallel plate test fixtures having a diameter of 25 mm. The thickness of the sample was 1mm and the instrument was operated in a steady shear mode at $25 \pm 1^\circ\text{C}$. Stress was increased linearly from 0 to 100 Pa over a period of five minutes, held at 100 Pa for 1 minute and then reduced linearly to zero during a five minute period. The maximum viscosity of the composition was measured at shear rates of 0.1 sec^{-1} and 1 sec^{-1} .
- [0079] The coefficient of thermal expansion of a cured composition was determined using a TA Instruments TMA 2940 thermomechanical analyzer. A sample having a diameter of 8 mm and a thickness of 1.9 mm was equilibrated at -155°C. and thereafter the temperature was increased at a rate of $5^\circ\text{C}/\text{min}$ to 200°C. The coefficient of thermal expansion was calculated by dividing the secant to the temperature versus dimensional change curve in the temperature range from -50 to -40°C. by the sample thickness.
- [0080] The storage modulus (G') of a cured composition was determined using a Rheometrics Dynamic Analyzer RDA II. The instrument was operated in the torsional rectangular bar mode with a frequency of 10 radians/s. A sample having a length of 36 mm, width of 12.7 mm, and thickness of 1.9 mm was held at -85°C. for twenty minutes and then increased at a rate of $3^\circ\text{C}/\text{min}$ to 50°C. Storage moduli were measured at -65 and 20°C.
- [0081] The tensile strength and elongation of a cured sample were determined in accordance with ASTM D 412C using a Monsanto T2000 tensiometer equipped with a type C die. Tensile strength and elongation measurements were carried out by taking the samples to ultimate failure.
- [0082] The hardness of a cured sample was measured using a Durometer Type A instrument. Three cured samples, each having a thickness of 1.9 mm, were stacked vertically and the hardness of the top exposed surface was measured.
- [0083] The peel adhesion of a cured composition on a polyimide substrate were measured using a Monsanto T2000 tensiometer. The curable composition was interposed between two square polyimide films, each having a thickness of 0.05 mm and width of 12.7 mm spaced 0.20 mm apart. The laminate was cured in an oven at 150°C. for 1 hour and then allowed to cool to room temperature. The force required to remove the polyimide films from the cured composition was measured at a peel angle of 180° and a peel rate of 5.1 cm per minute. The peel adhesion for one sample is an average of multiple readings taken during the course of one peel. The recorded values are the average peel adhesion of three identically prepared samples.
- [0084] Polymer Blend: a mixture consisting of 75 percent by weight of a dimethylvinylsiloxy-terminated polydimethylsiloxane having an average DP of 830 and a viscosity of $55 \text{ Pa} \cdot \text{s}$ at 25°C . and 25 percent by weight of a dimethylvinylsiloxy-terminated polydimethylsiloxane having an average DP of 434 and a viscosity of $2 \text{ Pa} \cdot \text{s}$ at 25°C .
- [0085] Resin: an organopolysiloxane having a Mn of 2,500 and consisting of $(\text{CH}_3)_2\text{CH}_2=\text{CHSiO}_{1/2}$ siloxane units, $(\text{CH}_3)_3\text{SiO}_{1/2}$ siloxane units and $\text{SiO}_{2/2}$ siloxane units, wherein the mole ratio of the combination of $(\text{CH}_3)_2\text{CH}_2=\text{CHSiO}_{1/2}$ units and $(\text{CH}_3)_3\text{SiO}_{1/2}$ units to $\text{SiO}_{2/2}$ units is 1.0, and the resin contains 6.0 mole percent of vinyl groups.
- [0086] Crosslinking Agent A: a trimethylsiloxy-terminated dimethylsiloxane-methylhydrogensiloxane having an average of five methylhydrogensiloxane units and three dimethylsiloxane units per molecule and a silicon-bonded hydrogen content of 0.7 to 0.8 percent by weight.
- [0087] Crosslinking Agent B: a trimethylsiloxy-terminated polymethylhydrogensiloxane having a silicon-bonded hydrogen content of 1.4 to 1.75 percent by weight and a viscosity of 0.22 to $0.04 \text{ Pa} \cdot \text{s}$.
- [0088] Adhesion Promoter: a blend consisting of 43.5 percent by weight of a hydroxyl-terminated dimethylmethylvinylsiloxane containing 2.25 to 4 percent by weight of silicon-bonded hydroxyl groups and 27.5 percent by weight of vinyl groups, 50 percent by weight of glycidoxypolytrimethoxysilane, and 6.5 percent by weight of a reaction product of the hydroxyl-terminated polymethylvinylsiloxane and glycidoxypolytrimethoxysilane.
- [0089] Inhibitor: 3,5-dimethyl-1-hexyn-3-ol.
- [0090] Catalyst: a platinum complex of 1,3-diethenyl-1,1,3,3-tetramethyldisiloxane.
- [0091] Pigment: carbon black.

Example 1

[0092] A silicone composition according to the present invention was prepared by mixing 100 parts of Polymer Blend, 100 parts of Resin, 19.0 parts of Crosslinking Agent A, 2.2 parts of Crosslinking Agent B, 4.8 parts of Adhesion Promoter, 2.4 parts of Pigment and 0.2 parts of Inhibitor. Catalyst (0.01 parts) was added to the mixture and stirring was continued for an additional 2 minutes. The viscosity of the composition is 18.1 Pa · s at a strain rate of 0.1 sec⁻¹ and 12.7 Pa · s at a strain rate of 1.0 sec⁻¹. The physical properties of the cured composition are shown in Table 1.

Comparative Example 1

[0093] A silicone composition was prepared using the method described in Example 1 and the following concentrations of components: 100 parts of Polymer Blend, 33.3 parts of Resin, 8.2 parts of Crosslinking Agent A, 0.95 parts of Crosslinking Agent B, 3.2 parts of Adhesion Promoter, 1.6 parts of Pigment, 0.16 parts of Inhibitor and 0.01 parts of Catalyst. The viscosity of the composition is 25.9 Pa · s at a shear rate of 0.1 sec⁻¹ and 20.3 Pa · s at a shear rate of 1.0 sec⁻¹. The physical properties of the cured composition are shown in Table 1.

Comparative Example 2

[0094] A silicone composition was prepared using the method described in Example 1 and the following concentrations of components: 100 parts of Polymer Blend, 66.7 parts of Resin, 13.6 parts of Crosslinking Agent A, 1.6 parts of Crosslinking Agent B, 4.0 parts of Adhesion Promoter, 2.0 parts of Pigment, 0.20 parts of Inhibitor and 0.01 parts of Catalyst. The viscosity of the composition is 17.2 Pa · s at a shear rate of 0.1 sec⁻¹ and 12.9 Pa · s at a shear rate of 1.0 sec⁻¹. The physical properties of the cured composition are shown in Table 1.

Comparative Example 3

[0095] A silicone composition absent Adhesion Promoter was prepared using the method described in Example 1 and the following concentrations of components: 100 parts of Polymer Blend, 100 parts of Resin, 15.1 parts of Crosslinking Agent A, 1.8 parts of Crosslinking Agent B, 2.4 parts of pigment, 0.24 parts of Inhibitor, and 0.01 parts of Catalyst. The viscosity of the composition is 15.6 Pa · s at a shear rate of 0.1 sec⁻¹ and 14.7 Pa · s at a shear rate of 1.0 sec⁻¹. The physical properties of the cured composition are shown in Table 1.

Example 2

[0096] A silicone composition identical to the composition in Example 1 was prepared, except 119 parts by weight of fused silica, having an average particle size of 4.5 ± 0.5 microns, was added to the composition after admixing the inhibitor. The viscosity of the composition is 64.6 Pa · s at a shear rate of 0.1 sec⁻¹ and 27.0 Pa · s at a shear rate of 1.0 sec⁻¹. The physical properties of the cured composition are shown in Table 1.

Comparative Example 4

[0097] A silicone composition identical to the composition in Example 2 was prepared, except the concentration of fused silica was 238 parts by weight. The viscosity of the composition is 379.3 Pa · s at a shear rate of 0.1 sec⁻¹ and 91.0 Pa · s at a shear rate of 1.0 sec⁻¹. The physical properties of the cured composition are shown in Table 1.

Table 1

Physical Properties	Examples		Comparative		Examples	
	1	2	1	2	3	4
Durometer Hardness (Shore A Scale)	75	83	41	61	73	89
Tensile Strength (MPa)	8.1	5.5	4.5	6.7	6.6	5.0
Elongation (%)	69	36	189	143	59	22
CTE ($\mu\text{m}/\text{m}^\circ\text{C}$ -50 to -40 $^\circ\text{C}$)	246	214	750	326	247	144
G' (MPa)						
-65 $^\circ\text{C}$	40.1	91.8	98.1	60	33.3	203
20 $^\circ\text{C}$	8.7	20.4	0.74	2.6	7.8	36.3
180° Peel Adhesion (N/m)	149 ^c	121 ^c	897 ^c	658 ^c	79 ^a	103 ^c

CTE = coefficient of thermal expansion,

G' = storage modulus,

a = adhesive failure, and

c = cohesive failure.

Claims

1. A silicone composition comprising:

(A) 100 parts by weight of a polydiorganosiloxane containing an average of at least two silicon-bonded alkenyl groups per molecule;

(B) 75 to 150 parts by weight of an organopolysiloxane resin having a number-average molecular weight of from 2,000 to 5,000 and comprising $\text{R}^3\text{R}^4\text{SiO}_{1/2}$ units and $\text{SiO}_{4/2}$ units wherein each R^3 is independently selected from monovalent hydrocarbon and monovalent halogenated hydrocarbon groups free of aliphatic unsaturation, R^4 is selected from R^3 and alkenyl groups, the molar ratio of $\text{R}^3\text{R}^4\text{SiO}_{1/2}$ units to $\text{SiO}_{4/2}$ units is from 0.6:1 to 1.1:1, and the resin contains an average of from 2.5 to 7.5 molar percent of alkenyl groups;

(C) an organohydrogenpolysiloxane having an average of at least three silicon-bonded hydrogen atoms per molecule in an amount to provide from one to three silicon-bonded hydrogen atoms per alkenyl group in components (A) and (B) combined;

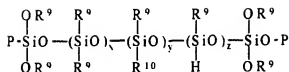
(D) an adhesion promoter in an amount to effect adhesion of the composition to a substrate; and

(E) a hydrosilylation catalyst in an amount to cure the composition.

2. The composition according to claim 1 wherein the polydiorganosiloxane (A) comprises a mixture of a first polydiorganosiloxane having a viscosity of from 45 to 65 $\text{Pa} \cdot \text{s}$ at 25 $^\circ\text{C}$. and a second polydiorganosiloxane having a viscosity of from 1.8 to 2.4 $\text{Pa} \cdot \text{s}$ at 25 $^\circ\text{C}$.

3. The composition according to claim 1 or 2, wherein the adhesion promoter (D) comprises a mixture of (i) a polysiloxane having an average of at least one silicon-bonded alkenyl group and an average of at least one silicon-bonded hydroxyl group per molecule and (ii) an epoxy-containing alkoxy silane.

4. The composition according to claim 1 or 2, wherein the adhesion promoter (D) is an organosilicon compound having the average formula:



wherein R^9 is a monovalent hydrocarbon group free of aliphatic unsaturation, R^{10} is an alkanyl group, x has a value of from 0 to 10, y has a value of from 0 to 10, z has a value of from 0 to 10, the sum $y + z$ is at least 1, and P has the formula:



wherein R^{11} is a covalent organic group.

5. The composition according to any of claims 1 to 4, further comprising an inorganic filler selected from alumina, boron nitride, aluminum nitride and fused silica.
6. The composition according to any of claims 1 and 3 to 5, wherein the polydiorganosiloxane A) has the general formula $\text{R}^2\text{R}^1_2\text{SiO}(\text{R}^1_2\text{SiO})_n\text{SiR}^1_2\text{R}^2$ wherein each R^1 is independently selected from monovalent hydrocarbon and monovalent halogenated hydrocarbon groups free of aliphatic unsaturation, R^2 is alkanyl, and n has a value such that the viscosity of the polydiorganosiloxane is from 0.5 to 100 Pa · s at 25°C.; the organopolysiloxane resin (B) comprises $(\text{CH}_3)_2\text{CH}_2=\text{CHSiO}_{1/2}$ units, $(\text{CH}_3)_3\text{SiO}_{1/2}$ units and $\text{SiO}_{4/2}$ units; the organohydrogenpolysiloxane (C) comprises a mixture of a trimethylsiloxy-terminated dimethylsiloxanemethylhydrogensiloxane and a trimethylsiloxy-terminated polymethylhydrogensiloxane; and the hydrosilylation catalyst (E) is a platinum catalyst.
7. A silicone elastomer obtained by a curing reaction of the composition of any of claims 1 to 6.
8. A silicone composition according to claim 1 in a multi-part container system where neither the polydiorganosiloxane nor the organopolysiloxane resin are present in the same part with the organohydrogenpolysiloxane and the hydrosilylation catalyst.
9. A method for preparing a silicone composition comprising mixing:
 - (A) 100 parts by weight of a polydiorganosiloxane containing an average of at least two silicon-bonded alkanyl groups per molecule;
 - (B) 75 to 150 parts by weight of an organopolysiloxane resin having a number-average molecular weight of from 2,000 to 5,000 and comprising $\text{R}^3\text{R}^4\text{SiO}_{1/2}$ units and $\text{SiO}_{4/2}$ units wherein each R^3 is independently selected from monovalent hydrocarbon and monovalent halogenated hydrocarbon groups free of aliphatic unsaturation, R^4 is selected from R^3 and alkanyl groups, the molar ratio of $\text{R}^3\text{R}^4\text{SiO}_{1/2}$ units to $\text{SiO}_{4/2}$ units is from 0.6:1 to 1.1:1 and the resin contains an average of from 2.5 to 7.5 molar percent of alkanyl groups;
 - (C) an organohydrogenpolysiloxane having an average of at least three silicon-bonded hydrogen atoms per molecule in an amount to provide from one to three silicon-bonded hydrogen atoms per alkanyl group in components (A) and (B) combined;
 - (D) an adhesion promoter in an amount to effect adhesion of the composition to a substrate; and
 - (E) a hydrosilylation catalyst in an amount to cure the composition.



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EUROPEAN SEARCH REPORT

Application Number
EP 99 30 8319

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
X	EP 0 835 918 A (DOW CORNING) 15 April 1998 (1998-04-15) * abstract * * Preparation Examples 2-4 * * page 2, line 23 - page 5, line 5; examples 24-33; table 9 * * page 2, line 48 - page 3, line 4 * -----	1-9	C08L83/04 C08L83/07 H01L23/29
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E	EP 0 955 328 A (DOW CORNING) 10 November 1999 (1999-11-10) * abstract * * paragraph '0006' - paragraph '0053'; claims; examples * -----	1-9	
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The present search report has been drawn up for all claims			
Place of search MUNICH		Date of completion of the search 21 January 2000	Examiner Hutton, D
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ANNEX TO THE EUROPEAN SEARCH REPORT
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